

### **the ultimate chemical equations pdf**

Answers for the questions in Chapter 8: ROUND 3 Exercise 8-1: Using the activity series, predict and balance the following single replacement reactions.

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Write a balanced molecular equation describing each of the following chemical reactions. (a) Solid calcium

carbonate is heated and decomposes to solid calcium oxide and carbon dioxide gas. (b) Gaseous butane,  $C_4H_{10}$ , reacts with diatomic oxygen gas to yield gaseous carbon dioxide and water vapor.

#### **4.1 Writing and Balancing Chemical Equations**

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